TABLE III CHLOROFORM-INDUCED CARDIAC ARRHYTHMIAS AFTER TREATMENT WITH 4 COMMON. CLINICALLY USED ANTIARRHYTHMIC AGENTS

COMMON, CEINICALET COMP IN TRAMMITTING TOEMAS											
Drug	Number of mice	Equimolar dose (mg/kg)	Average rate (beats/min \pm S.E.)	% protected	% toxic	'% fatal					
Quinidine∙SO₄	40	118.47	129.05 ± 13.42	83.3	0.0	0.0					
dl -Propranolol \cdot HCl	40	77.47	185.61 ± 12.45	6 0.0	46 .0	0.0					
$\mathbf{Procainamide} \cdot \mathbf{HCl}$	40	58.92	291.57 ± 17.11	13.3	0.0	0.0					
$Lidocaine \cdot HCl$	40	70.58	182.77 ± 14.78	56.7	86.7	0.0					

structural requirements for substitution on N is unclear since good protection resulted from compounds having primary, secondary, and tertiary amine groups. In almost all cases, the primary amine derivatives caused greatly reduced toxicity when compared with the analogous substituted compounds (see 5, 6, 11, 14). In the comparison of the secondary amine analogs, an increase in the lipid solubility or hydrophobic character of the substituent increased both the antifibrillatory protection and the toxicity (see 3, 4). The substitution of more hydrophobic groups at any position on the saturated ring increased the antifibrillatory activity and also the toxicity (see 6, 7, and 9–17). This may be due to an enhanced transport capability related to the relative partition characteristics of the molecule, as described by Levy,⁵ rather than any specific receptor requirement.

From the results it is apparent that some of the 2aminotetralins tested possess very good antifibrillatory protection properties compared with the commonly used antiarrhythmic agents and show little or no acute toxicity in effective doses. The relationship between toxicity and antiarrhythmic efficacy is very important. Although many agents possessing relatively potent antiarrhythmic activity have been developed, a great number of them have exhibited untoward toxic properties. Thus the absence of noticeable toxicity in dosage levels resulting in good antiarrhythmic activity in the screening experiments in highly desirable.⁶

Experimental Section

Adult male mice (Carworth strain CF 1) (22-28 g were weighed and injected ip with the test compound and placed in separate glass containers. The injections were given to groups of animals with 2-min intervals between individual animal treatments. After injection each animal was observed for toxic symptoms. Exactly 10 min after treatment each animal was transferred to a 250-ml beaker that held a cotton pad saturated with 20 ml of CHCl₂. The animal was removed immediately after respiratory arrest and the heart was quickly exposed by removing the anterior thoracic wall without touching the heart with the surgical instruments. The heart rate and fibrillatory movements were recorded with the aid of a stop watch and a binocular dissecting microscope. Any animal showing fibrillation or ventricular rates in excess of 200 beats/min was defined as unprotected. All animals exhibiting rates below 200 beats/min were reported as protected from fibrillation. In all cases where sufficient drug was available groups of 40 mice were used. Values for the percentage of animals protected were used to calculate an ED₅₀ according to Litchfield and Wilcoxon.⁷ All other statistical calculations were performed according to Steel and Torrie.⁸

Antimalarial and Other Biological Activities of Some 2'-Alkyl and 2'-Aryl Derivatives of Cinchona Alkaloids¹

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We report an investigation undertaken primarily to compare with activities of 2'-alkyl and 2'-aryl derivatives of quinine, quinidine, and their 10,11-dihydrides against experimental malarias caused by Plasmodium berghei and P. gallinaceum in mice and chicks, respectively, in the framework of studying the effect of blocking the 2 position of the quinoline nucleus.²⁻⁴ The substances were also examined for cardiac antiarrythmic and antibacterial activity.

The 2'-alkyl and 2'-aryl derivatives were prepared from the alkaloid ar-N-oxides and appropriate Grignard or Li organometallic reagents (cf. ref 5). Satisfactory yields were obtained with simple primary alkyl Grignard reagents. *i*-BuMgBr and sec-RMgBr yielded predominantly the parent diamine along with a low yield of the corresponding 2'-alkyl derivative. The reaction failed with t-BuMgBr. However, 2'-t-Bu and 2'-cyclopropylquinidine were obtained from quinidine ar-Noxide and the corresponding Li alkyl. The 10,11dihydroquinidines and dihydroquinines were obtained by catalytic hydrogenation of the appropriate quinidine and quinine derivative I and II, respectively. Under the conditions used, no loss of halogen from the 2'-halophenyl derivatives was observed. Physical constants for the new compounds and other data are listed in Tables I-IV.

Antimalarial Activity --- Activity was measured against P. berghei in mice and P. gallinaceum in chicks by previously described methods.⁶ The data in Table I show that 2'-alkyl substituents lower the maximum tolerated dose (MTD) thereby restricting testing to low dosage levels. Also, 2'-aryl substituents confer in-

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⁽⁶⁾ T. S. Osdene, P. B. Russell, and L. Rane, J. Med. Chem., 10, 431 (1967).

			_		STITUTED QUINIDINES				
Compd ^a	R	Mp, °C	Recrystn solvent ^b	Yield, %	Formula	Analysis	Activity index ^c	P. berghei	P. gallinaceum
1	CH_3	160	А	75	$C_{21}H_{26}N_2O_2$	C, H, ^d N	Ι	320	120
-	0110	100		••	02111201(202	0, 11, 11	ĪI	9.4	Cure
							III	Ne ⁷	60
1a		202 - 218	В		$C_{21}H_{26}N_2O_2\cdot 2HCl\cdot 1/_4H_2O$	C, H, Cl, ^g N	IV	160	60
2	C_2H_5	138 - 140	С	39	$C_{22}H_{28}N_2O_2$	C, H, N	Ι	40	Ne
							II	Ne	Ne
							III		60
2a		171 - 175	D, E		$C_{22}H_{28}N_2O_2\cdot 2HCl$		\mathbf{IV}		Ne
3	$C_{3}H_{7}$	Amorph		33	$C_{23}H_{30}N_2O_2$		Ι	80	Nt^h
							II	0	
3a		175	D, E		$\mathrm{C}_{23}\mathrm{H}_{30}\mathrm{N}_{2}\mathrm{O}_{2}\cdot\mathrm{2HCl}\cdot\mathrm{H}_{2}\mathrm{O}$	C, H, Cl, N			
4	$C(CH_3)_3$	\mathbf{Amorph}			$C_{24}H_{32}N_2O_2{}^{i}{}^{,j}$			\mathbf{Nt}	\mathbf{Nt}
4a		240	D, E	11	$C_{24}H_{32}N_2O_2\cdot 2HCl$				
5	$(\mathrm{CH}_2)_2\mathrm{CH}(\mathrm{CH}_3)_2$	\mathbf{Amorph}		17	$\mathrm{C}_{25}\mathrm{H}_{34}\mathrm{N}_{2}\mathrm{O}_{2}$		Ι	320	\mathbf{Nt}
							II	0	
5a		170 - 174	D, E		$C_{25}H_{34}N_2O_2\cdot 2HCl\cdot^2/_3H_2O$	C, H, Cl, N			
6	\neg	59 - 72	\mathbf{F}	14	${ m C_{23}H_{28}N_2O_2}^k$			\mathbf{Nt}	\mathbf{Nt}
6a		174	D, E		$C_{23}H_{28}N_2O_2\cdot 2HCl$				
7	$(\mathrm{CH}_2)_{3}\mathrm{N}(\mathrm{CH}_3)_{2}$	\mathbf{Amorph}		19	$C_{25}H_{35}N_{3}O_{2}{}^{l}$				
7a		\mathbf{Amorph}			$C_{25}H_{35}N_3O_2\cdot 3HCl$			\mathbf{Nt}	\mathbf{Nt}
8	C_6H_5	193 - 196	A, \mathbf{E}	37	$C_{26}H_{28}N_2O_2$	С, Н, N	I	640	Ne
							II	Cure	
							III	320	15
8a		\mathbf{Amorph}			$\mathrm{C}_{26}\mathrm{H}_{28}\mathrm{N}_{2}\mathrm{O}_{2}\cdot\mathrm{2HCl}$		\mathbf{IV}	160	\mathbf{Ne}
9	$4-ClC_6H_4$	118 - 122	\mathbf{E}	23	$C_{26}H_{27}ClN_2O_2\cdot {}^1/_2H_2O$	C, H, Cl, N	Ι	>640	>240
							II	Cure	Cure
							III	320	<60
9a	-	187 - 191	D, E		$C_{26}H_{27}ClN_2O_2\cdot 2HCl$		IV	40	Ne
10	$4-CF_{3}C_{6}H_{4}$	$\mathbf{A}\mathbf{morph}$		40	$C_{27}H_{27}F_{3}N_{2}O_{2}$		Ι	>640	>640
							II	Cure	14.7
			_				III	160	
10a		198 - 203	A, \mathbf{E}		$\mathrm{C_{27}H_{27}F_{3}N_{2}O_{2}\cdot 2HCl\cdot H_{2}O}$	C, H, Cl, N	IV	20	<20
a = h	ydrochloride, all anti	inialarial tes	t results w	ere obt	ained with HCl salts. $^{b}A =$	MeOH; $B = 1$	Me ₂ CO;	$C = CH_{a}$	$(CH_2)_5CH_3;$
$D = i - Pr^{\dagger}$	$OH; E = Et_2O; F =$	$= CH_3(CH_2)_4$	CH ₃ ; G =	· CH ₂ C	H_2 ; $H = aq CH_3O(CH_2)_2OCH_3O(CH_2)O(CH_2)O(CH_$	l_3 . $c_1 = Max t_1$	olerated	dose, MT	$D \ln mg/kg$

TABLE I 2'-Substituted Quinidines

^a a = hydrochloride, all antimalarial test results were obtained with HCl salts. ^b A = MeOH; B = Me₂CO; C = CH₃(CH₂)₅CH₃; D = *i*-PrOH; E = Et₂O; F = CH₃(CH₂)₄CH₃; G = CH₂Cl₂; H = aq CH₃O(CH₂)₂OCH₃. ^c I = Max tolerated dose, MTD in mg/kg (MTD = dose at which no toxic death occurred). II = Increase in mean survival time at MTD, in days; III = Minimum dose giving cure in mg/kg; IV = Minimum dose showing activity. ^d H: Calcd 7.74, found 7.25; ^e Cure is defined as a survival time of 60 days of the treated mouse over its control and of 30 days of the treated chick over its control. ^f Ne = not established. ^o CI: Calcd 17.05, found 15.07. ^h Nt = not tested. ⁱ The molecular composition of difficult to analyze products was determined on an AEI MS-902 mass spectrometer at low resolution. The spectra were examined for M⁺ and major fragmentation; only M⁺ is reported. ^j m/e: Calcd 380, found 380. ^k m/e: Calcd 364, found 364; ⁱ m/e: Calcd 409, found 409.

				TABLE II				
			2'-Subs	STITUTED 10,11-DIHYDRO	QUINIDINES			
Compd	R ^a	Mp, °C	Recrystn solvent ^b	Formula	Analysis	Activity index ^c	P. berghei	P. gallinaceum
11 11a	$C_{3}H_{7}$	168-171 184-186	Е D, Е	${f C_{23}H_{32}N_2O_2}\ {f C_{23}H_{32}N_2O_2\cdot HCl}$	С, Н, N	Ι	20	Nt ^a
12	$\mathrm{C}_{6}\mathrm{H}_{5}$	260-270	Α, Ε	$\mathrm{C_{26}H_{\$0}N_2O_2\cdot 2H_2O}$	C, H, N	I II III	$>640 \\ 10.6$	>120 Cure [•] Ne ⁷
12a 13	4-ClC ₆ H ₄	184–187 163–174	D, E E	$\begin{array}{c} C_{26}H_{30}N_{2}O_{2}\cdot 2HCl\\ C_{26}H_{29}ClN_{2}O_{2}\end{array}$	C, H, Cl, N	IV I II III	320 >640 Cure 320	Ne >120 Cure 60
13a 14 14a	4-CF ₃ C ₆ H ₄	185–189 Amorph 201–205	D, E A, E	$\begin{array}{c} C_{26}H_{29}ClN_2O_2\cdot 2HCl\\ C_{27}H_{29}F_8N_2O_2\\ C_{27}H_{29}F_8N_2O_2\cdot 2HCl\\ 1.5H_2O \end{array}$	C, H,º Cl, N	ΙΫ	40 Nt	30Nt

 $^{a-f}$ See footnotes a, b, c, h, e, f, respectively, in Table I. ^g H: Calcd 5.43, found 4.91.

creased potency thereby enabling cures' to be observed. No significant differences in activity were detected between the quinine and quinidine derivatives and their corresponding dihydrides. In the 2'-aryl quinine series the minimal active dose decreased (*i.e.*, anti-

(7) See footnote e in Table I.

malarial potency increased) in the following order of 2' substituents: $C_6H_5 > p-FC_6H_4 > p-CH_3OC_6H_4 > p-CIC_6H_4 > p-CF_3C_6H_4$. In our hands, 2-p-trifluoromethylphenylquinine, showing activity at 10 mg/kg, appears some 32 times as potent as quinine itself. Unfortunately, all of the 2'-arylquinines and quinidines

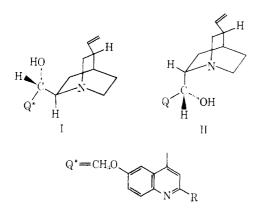
2'-SUBSTITUTED QUININES										
Compd	\mathbb{R}^{a}	Mp, °C	Recrystn solvent ^b	Yield, %	Formula	Analysis	Activity index ^c	P. berghei	P. gallinaceum	
15	$C_6 H_5{}^d$	151	B, E	44	$\mathbf{C_{26}H_{28}N_2O_2}$		Ι	>640	>120	
							II	9.6	Cure	
							III		60	
15a		215	В, G		$C_{26}H_{28}N_2O_2\cdot 2HCl$		IV	320	30	
16	FC ₆ H ₄	130	\mathbf{E}	45	$\mathrm{C}_{26}\mathrm{H}_{27}\mathrm{FN}_{2}\mathrm{O}_{2}$		Ι	640	120	
							II	\mathbf{Cure}	Cure	
							III	640	120	
16a		215	A, E		$C_{26}H_{27}FN_2O_2\cdot 2HCl$	C, H, F, Cl, ¹ N	\mathbf{IV}	160	30	
17 4-CH ₃ OC	$4-CH_{3}OC_{6}H_{4}$	4			$C_{27}H_{30}N_2O_3$		I	>640	>120	
							11	Cure	\mathbf{Cure}	
							III	160	30	
17a		215	Н	35	$C_{27}H_{30}N_2O_3\cdot 2HCl$	C, H, Cl, N	IV	80	<30	
18	4-ClC ₆ H₄				$\mathrm{C}_{26}\mathrm{H}_{27}\mathrm{ClN}_{2}\mathrm{O}_{2}$		Ι	>640	120	
							II	Cure	Cure	
							III	80	60	
18a		215	A, \mathbf{E}	48	$C_{26}H_{27}ClN_2O_2\cdot 2HCl$	C, H, Cl, ^{<i>g</i>} N	IV	80	30	
19	$4-CF_3C_6H_4$				$C_{27}H_{27}F_3N_2O_2$		I	>640	>120	
							II	Cure	10.6	
							III	20		
19a		210 - 218	А, Е	32	$C_{27}H_{27}F_{3}N_{2}O_{2}\cdot 2HCl$	C, H, ^{h} F, ^{i} Cl, N	\mathbf{IV}	10	Ne^{i}	

TABLE III

^{a-c} See footnotes *a-c*, respectively, in Table I. ^d Kobayshi, Yakugaku Zasshi, **71**, 260 (1951), reported mp 145–147°. ^e See footnote *e* in Table I. ^f Cl: Calcd 14.4, found 13.8. ^e Cl: Calcd 20.94, found 20.20. ^h H: Calcd 5.58, found 5.07. ^f F: Calcd 10.18, found 9.55. ^f See footnote ^f in Table I.

$\mathbf{T}_{\mathbf{ABLE}}$ IV								
2'-Substituted	10,11-Dihydroquinines							

Compd^a	R	Mp, °C	Recrystn solvent ^b	Yield, %	Formula	Activity index ^c	P. berghei	P. gallinaceum				
20	$4-CF_{3}C_{6}H_{4}$				$C_{27}H_{29}F_{3}N_{2}O_{2}{}^{d}$	I	>320	>120				
						III	Cure^{e}	10.8				
						III	80					
20a		\mathbf{Amorph}	M, E	50	$\mathrm{C}_{27}\mathrm{H}_{29}\mathrm{F}_{3}\mathrm{N}_{2}\mathrm{O}_{2}\cdot\mathrm{HCl}$	IV	10	120				
a - c See f	ootnotes $a-c$ in Ta	able I. $d m/e$:	Calcd 470, for	ınd 470.	^{e} See footnote e in Table I.							



showed severe phototoxicity⁸ in a laboratory animal test.⁹ Quinine itself is inactive in this test while displaying slight photoactivity in an *in vitro* screening procedure.¹⁰ The phototoxicity of 6,8-dichloro-2-phenyl- α -phenyl- α -2-piperidyl-4-quinolinemethanol, the potent antimalarial arylquinoline closely related to series I and II, has produced severe complications in man,¹¹ and it may be that, in animals, the phototoxicity of these compounds is directly related to their antimalarial potencies.

(8) Private communication by Dr. D. P. Jacobus and Col. William E. Rothe, Walter Reed Army Medical Center, Washington, D. C.

(9) W. E. Rothe, and D. P. Jacobus, J. Med. Chem., 11, 366 (1968).
(10) I. G. Fels, *ibid.*, 11, 887 (1968).

(11) T. N. Pullman, L. Eichelberger, A. S. Alving, R. Jones, Jr., B. Craige, Jr., and C. M. Wherton, J. Clin. Invest., 27, 12 (1948).

TABLE V ACTIVITY OF STANDARD COMPOUNDS AGAINST P. berghei in Mice and P. gallinaceum in Chicks

Standard compd	Activity index ^a	P. berghei	P. gallinaceum
Quinidine	Ι	>640	>320
	II	11.7	Not active
	\mathbf{IV}	320	
Dihydroquinidine	I	>640	Ne
	II	11.7	Not active
	IV	640	
Quinine	I	>320	>160
	II	13.2	3.2
	\mathbf{IV}	320	160
Dihydroquinine	Ι	>640	>120
	II	6.5	3.2
	\mathbf{IV}	640	Ne^{b}

^a See footnote c in Table I, the compounds were tested as the free base. ^b See footnote f in Table I.

Antiarrythmic and Antibacterial Activity.—The 2'alkylquinidines 1, 2, 6, and 7, largely retained quinidine-like activity in a test recording the effective refractive period of the isolated guinea pig atria,¹² the first showing a potency equal to or greater than that of quinidine.

All of the new substances reported here were also screened against strains of bacteria noted in Table VI. In the quinidine series antibacterial activity ran closely

(12) G.S. Dawes, Brit. J. Pharmacol., 1, 90 (1946).

ANTIBACTERIAL ACTIVITY OF 2'-ALKYL AND 2'-ARYL DERIVATIVES OF CINCHONA ALKALOIDS													
Test $Organism^a$	$1a^b$	2a	6	7a	8a	9a	10a	14 a	15a	16a	17a	18a	19a
Bacillus subtilis 6633	Na	250°	125	Na	15.6	7.81	7.81	1.95	15.6	7.81	3.90	15.6	7.81
Staphylococcus aureus 6538P	Na	Na	250	Na	62.5	15.6	7.81	3.90	62.5	31.3	15.6	62.5	15.6
S. aureus Smith	Na	Na	250	Na	62.5	15.6	7.81	3.90	62.5	31.3	15.6	62.5	15.6
S. aureus CHP	Na	Na	250	Na	62.5	7.81	7.81	1.95	62.5	15.6	15.6	15.6	7.81
S. aureus 53-180	Na	Na	250	Na	62.5	15.6	Na	1.95	62.5	15.6	15.6	62.5	15.6
$My cobacterium \ smegmatis$													
10143	Na	Na	Na	\mathbf{Na}	Na	Na	7.81	Na	Na	Na	Na	Na	Na
Neisseria catarrhalis 8193	Na	Na	250	Na	31.3	15.6	Na	1.95	31.3	15.6	7.81	31.3	7.81
Pseudomonas aeruginosa													
10145	Na	Na	Na	$\mathbf{N}\mathbf{a}$	\mathbf{Na}	Na	Na	Na	Na	Na	Na	Na	Na
Escherichia coli 6880	\mathbf{Na}	Na	250	Na	Na	Na	Na	Na	Na	Na	Na	Na	Na
E. intermedia 65-1	Na	Na	Na	Na	\mathbf{Na}	Na	Na	Na	Na	Na	Na	Na	Na
Salmonella paratyphi 11737	\mathbf{Na}	Na	Na	Na	Na	Na	Na	Na	Na	Na	Na	Na	Na
Enterobacter aerogenes 884	Na	Na	250	Na	Na	62.5	62.5	62.5	31.3	31.3	15.6	62.5	62.5
Klebsiella pneumoniae 10031	Na	Na	250	Na	62.5	7.81	15.6	7.81	31.3	31.3	15.6	62.5	62.5
Bordetella bronchiseptica 4617	Na	Na	250	Na	250	125	31.3	62.5	62.5	62.5	62.5	Na	62.5
Proteus vulgaris 6896	Na	Na	Na	Na	Na	Na	Na	Na	Na	Na	Na	Na	Na
Herellea sp. 9955	Na	Na	250	Na	Na	Na	$\mathbf{N}\mathbf{a}$	Na	250	250	250	Na	Na
^a The antibacterial activity was tested by the Agar Serial Dilution Technique. ^b For numbering of compound see Tables I-IV:													

TABLE VI ANTIPACTERIAL ACTIVITY OF 2'-ALKYL AND 2'-ARYL DERIVATIVES OF CINCHONA ALKALOIDS

^a The antibacterial activity was tested by the Agar Serial Dilution Technique. ^b For numbering of compound see Tables I-IV; ^c Minimal inhibitory concentration in $\mu g/ml$.

parallel to the observed antimalarial activity, the 2'alkyl derivatives showing little or no activity, and the 2'-aryl derivatives being moderately to highly active, with 2'-p-trifluoromethylphenyl member again the most potent of the series.

Experimental Section

Melting points were taken on a Kofler block under microscopic magnification or in capillary tubes using the Thomas-Hoover apparatus and are uncorrected. Pmr spectra of all compounds were measured with a Varian Associates A-60 spectrometer on 10-15% solution in CDCl₃ or DMSO-d₆ and confirmed the suggested structures. Compounds were examined for purity by tlc on silica gel plates irrigated with the lower phase of the equilibrated system CHCl₃-MeOH-NH₄OH (9:1:5), visualization was carried out with Dragendorff reagent.¹³ Quinine and quinidine ar-N-oxides were prepared by Ishikawa's method.¹⁴

Materials.—Commercially available Grignard reagents were used whenever possible, MeMgI, EtMgBr, PhMgBr were obtained from Arapahoe Chemical Company, Boulder, Colo. Others were prepared *in situ* from their respective halides and magnesium.

2'-Methylquinidine.—Quinidine ar-N-oxide (2.0 g, 5.9 mmoles) in 40 ml of anhyd C₆H₆ was added dropwise to a stirred 3 M sol of MeMgI in Et₂O (25 ml, 75 mmoles) over a period of 15 min. The mixture was refluxed under N₂ for 30 min then cooled to -10° and treated with ice water followed by excess 2 N HCl. The aq phase was sepd washed with Et₂O and strongly basified with ice-cold KOH soln CHCl₃ (250 ml) was added and the mixture stirred vigorously for a few minutes and then filtered through Celite. After washing the filter aid with CHCl₃ the extracts were combined washed with brine and dried (K₂CO₃). Evapn of the CHCl₃ gave 2.1 g of residue, readily crystallizing on trituration with MeOH to give 1.5 g of product, (as the MeOH) solvate), mp 150°. The analytical sample (from MeOH), mp 160°, lost MeOH of solvation after drying for 5 hr at 80° *in vacuo*. **2'-Cyclopropylquinidine.**—To a soln of cyclopropyllithium in Et₂O (250 ml), prepared from Li (4.4 g, 646 mmoles) and cyclopropyl bromide (36.7 g, 331 mmoles)¹⁵ was added while stirring at 0-5° under N₂ a soln of quinidine ar-N-oxide (10.0 g, 29.5 mmoles) in C₆H₆ (200 ml). When addition was complete stirring was continued while allowing the reaction to warm to room temp over a period of 1 hr. The mixture was then chilled in an ice bath and 2 N HCl was added. The layers were sepd and the aq phase washed with a small quantity of Et₂O. It was then basified with cold 50% NaOH soln, and extd with 5 portions of CHCl₃. The exts (CHCl₃) were combined, dried, and taken to dryness. The oily residue was chromatographed on a column of Florisil, fractions eluted with 1 and 2% MeOH in CHCl₃ crystd from hexane to give 1.5 g of product.

2'-(p-Chlorophenyl)dihydroquinidine.—2'-(p-Chlorophenyl)quinidine ·2HCl (2.6 g, 5 mmoles) was dissolved in MeOH (100 ml) and added to prereduced 5% Pd-C (0.89 g) in 25 ml of MeOH and hydrogenated at atmospheric pressure for 45 min until 5 mmoles of H₂ had been taken up. The catalyst was then filtered off and the filtrate taken to dryness. The residue was dissolved in H₂O and basified by the addition of NaOH soln. The aq phase was then extracted with CHCl₃, the combined extracts were dried and coned *in vacuo*. The residue gave 1.02 g of cryst product (from Et₂O).

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